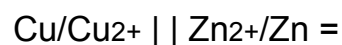
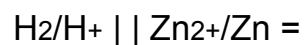
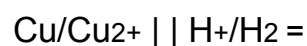
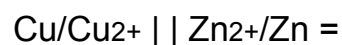
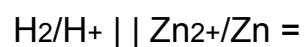
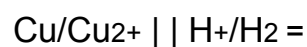


1. Calculate E_{cell} for each voltaic cell tested.



2. For each of the voltaic cells tested, write out the two redox half-reactions (oxidation and reduction).



3. Use half redox half-reactions to write the balanced net ionic equation for each voltaic cell.

